

Science 10 Nomenclature Review

Name: _____

Write the Formula or Name

A. Simple Compounds

1. NaCl _____

2. Calcium oxide _____

3. Barium chloride _____

4. BeBr₂ _____

5. Mg₃P₂ _____

B. Compound with polyatomic ions

6. Sodium hydroxide _____

7. BaSO₄ _____

8. Ammonium nitrate _____

9. Ammonium dichromate _____

10. Calcium phosphate _____

C. Transition Metals

11. Mercury II Oxide _____

12. NiCO₃ _____

13. Mn(OH)₄ _____

14. Tin IV acetate _____

15. Copper I sulphide _____

D. Covalent (molecular) compounds)

16. Carbon monoxide _____

17. Carbon tetrachloride _____

18. NO₃ _____

19. SCl₂ _____

20. Dinitrogen pentaphosphide _____

Types of Reactions practice

Match the correct equation to type of reaction

___ Single Replacement

___ Double replacement

___ Synthesis

___ Decomposition

___ Neutralization

